Math 430 * Math 635 * Fall 2015 * Victor Matveev

Derivation of the Nernst Equation for ion "X":

Let's think of a pore across the membrane as a one-dimensional space, and direct the "x"-axis across the pore (into or out of the cell). Consider an ion "X" of charge q moving within this pore along axis x.

At reversal potential, the diffusive current of ions precisely balances their field-driven current:

 $J_D + J_F = 0 \leftarrow$ These are current densities in units of # / (s m²)

Diffusive current is given by the Fick's law of diffusion (ions move against concentration gradient):

$$J_{D} = -D\frac{d[X]}{dx} = -u \ k_{B}T\frac{d[X]}{dx}$$

Here we used Einstein's relationship between diffusion coefficient and temperature: $D = u k_B T$

Parameter *u* is the mobility, which is the proportionality constant between equilibrium ("terminal") velocity and force in viscous flow: v = u Force

Extra curious fact (Stokes-Einstein-Sutherland equation): $u = 1/[6\pi (radius)(viscosity)]$

Now, recall that electrostatic force (in 1D) equals Force = $q\mathbf{E} = -q\nabla V = -q\frac{dV}{dx}$

Note that the product between velocity and density gives current density (flux): $\frac{m}{s}\frac{\#}{m^3} = \frac{\#}{m^2s}$

Therefore, for the electric field-driven flux, we have to multiply velocity and density, yielding

$$J_E = v[X] = [X] \cdot u \underbrace{\text{Force}}_{-a \ dV/dx} = -u \ q \ [X] \frac{dV}{dx}$$

We can finally examine the equality between the two current densities:

$$J_E = -J_D \implies -u q [X] \frac{dV}{dx} = u k_B T \frac{d[X]}{dx}$$

$$\Rightarrow \quad \frac{dV}{dx} = -\frac{k_B T}{q} \frac{1}{[X]} \frac{d[X]}{dx} = -\frac{k_B T}{q} \frac{d \ln[X]}{dx}$$

Now integrate from the outside to the inside of the membrane:

$$V = V_{in} - V_{out} = -\frac{k_B T}{q} \left(\ln \left[X \right]_{in} - \ln \left[X \right]_{out} \right) = -\frac{k_B T}{q} \ln \frac{\left[X \right]_{in}}{\left[X \right]_{out}} = \left| \frac{k_B T}{q} \ln \frac{\left[X \right]_{out}}{\left[X \right]_{in}} \right|_{in}$$

Note that the ion charge q is the product of its valence z and the elementary charge e