Metal-insulator transition in CuIr₂S₄: XAS results on the electronic structure

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S K and Ir L_3 x-ray absorption measurements across the temperature-induced metal (*M*) to insulator (*I*) transition in CuIr₂S₄ are presented. Dramatic S K-edge changes reflect the Ir *d*-electronic state redistribution across this transition. These changes, along with a detailed consideration of the *I*-phase structure, motivate a model in which the *I*-phase stabilization involves an interplay of charge and *d*-orbital orientation ordering along Ir chains, a quadrupling of the Ir-chain repeat unit, and correlated dimer spin-singlet formation.

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Metal-insulator transitions involving transition metal (*T*) compounds have been of intense interest in recent years on both fundamental and technological grounds.^{1,2} This field has been dominated by 3d-row *T*-oxide compounds, by virtue of the renaissance in these materials that followed the discovery of high- T_c superconductivity.³ The subclass of mixed valent metallic compounds that "charge order" into an insulating state has been the focus of special recent interest.² The compound CuIr₂S₄ stands out as unique in this subclass for a number of reasons.^{4–9}

The spinel structure CuIr₂S₄ compound has a paramagnetic, high-temperature, homogeneously mixed Ir³⁺/Ir⁴⁺, metallic (M) phase, which undergoes a first-order transition (near 230 K) to a low-temperature, charge-ordered, diamagnetic, insulating (I) phase.⁴⁻⁹ Recent definitive x-ray and neutron scattering measurements showed this I phase to involve a complex ordering of Ir⁴⁺-spin-singlet dimers and undimerized Ir³⁺ sites.¹⁰ This finding is novel, since such dimerization transitions have previously occurred almost exclusively in compounds with clearly defined quasi-onedimensional (1D) chains, and certainly not in a complex three-dimensional (3D) type structure.¹¹ Moreover, the complex pairwise ordering of Ir³⁺-Ir³⁺ nondimer and Ir⁴⁺-Ir⁴⁺ singlet-dimer moieties appears not to have been previously observed. In addition, the broadness of the 5d orbitals makes such M-I transitions rare in 5d-row compounds, and still rarer in a nonoxide (i.e., S) 5d compound. These properties, along with the previous dearth of precise structural or electronic information, have hindered the development of the outlines of a theory for the M-I transition in CuIr₂S₄. In this paper, we present S and Ir XAS results spanning the M-I transition in $CuIr_2S_4$. The S K-edge results, along with a careful review of the dimerized/charge-ordered chain character of the I phase, motivate a proposal for a specific electronic/structural basis from which to approach understanding of this novel *I-M* transition.

The sample preparation and characterization techniques were as discussed in Ref. 10. The S K and Ir L_3 XAS measurements were, respectively, performed on beamlines X19A and X18B at the Brookhaven National Synchrotron Light Source, using methods discussed in Refs. 12 and 13. The low-temperature XAS measurements on X19A utilizes a nitrogen cryostat (in the fluorescence mode), and those on

X18B used a displex refrigerator (using the transmission mode).

Previous photoemission spectroscopy measurements on CuIr_2S_4 showed a subtle loss of electronic states at 0.2 eV below the Fermi energy (E_F) in the *I* phase.⁶ These photoemission and *I*-phase inverse photoemission results manifested substantial departures⁶ from band-structure predictions⁹ on a wider energy scale. Thus the generic gap formation below E_F , and the absence of any information about above- E_F density-of-state (DOS) *changes* across this *M-I* transition, have provided little guidance to direct the theoretical attack on this interesting system.

XAS has been important in elucidating atomic/orbitalspecific electronic structure in many classes of transition metal compounds. The combination of O K-edge (probing O p states) and Cu $L_{2,3}$ -edge (probing Cu d states) measurements emphasized the crucial importance of hybridized O p/Cu d hole states in the physics of high- T_c cuprates.¹⁴ Importantly, O K-edge measurements on VO2 clarified the specific electronic structure changes behind its metal to dimerized-insulator transition.¹⁵ Systematic O K-edge measurements in T(3d)-O compounds have profiled (via their threshold structure) the important variations in their hybridized O p/T d empty states.¹⁶ 4d/5d-hole states have been studied extensively by T $L_{2,3}$ -edge spectroscopy in transition-metal compounds.¹² Finally, T(3d) K-edge XAS has been used to chronicle the valence variations in manganites and the charge donation to Cu in electron-doped high- T_c materials.¹³ This background motivates our S K- and Ir L_3 -edge studies across the *M-I* phase transition in CuIr₂S₄. Cu K-edge measurements in our laboratory, along with previous NMR⁵ and band-structure⁹ calculations, indicate a d^{10} Cu^{1+} state in $CuIr_2S_4$, allowing us to neglect Cu d state influences on the S states above E_F .

The elemental S *K* edge in Fig. 1(a) is dominated by an intense "white line" (WL) feature, due to dipole transitions into empty 3p states. In transition-metal sulfide compounds, one typically observes¹⁷ a diminution of the WL intensity due to T to S charge transfer and the appearance of prominent threshold features, shifted down by 0–5 eV from the elemental-S WL, due to hybridized S p/T d states. The S *K* threshold features probe the T(*d*) DOS (weighted by transi-



FIG. 1. (a) The S K edges of elemental S and CuT_2S_4 , with T = Cr, Co, and Ir. The a'- and a-threshold features are, respectively, associated with T $d t_{2g}$ and e_g state hybridization. (b) The S K edges of CuIr_2S_4 , at temperatures T=195 K (in the I phase) and 300 K (in the M phase). (c) The Ir L_3 edges of elemental Ir, IrO₂, CuIr_2S_4 , and CuIr_2S_4 . The intensity in the A and A' positions are associated with Ir $d t_{2g}$ and e_g final states, respectively.

tion matrix element effects) in sulfides in the same sense as O K threshold features do in oxides.¹⁶ The S K-edge spectra for the spinel compounds CuT_2S_4 (T = Cr, Co, and Ir) in Fig. 1(a) illustrate this for the octahedral ligand field case, where the d orbitals are split into a lower t_{2g} sextet and an upper e_g quartet. For the isoelectronic 3d Co and 5d Ir compounds, the empty states are $t_{2g}^{0.5} - e_g^4$ and for the Cr compound they are $t_{2g}^{2.5} - e_g^4$. The *a* and *a*' features in the S K spectra are associated with the empty S p states, hybridized with the empty t_{2g} and e_g states, respectively. Consistent with band calculations,⁹ the ligand field splitting for T=Ir is large (yielding a resolved a' - a splitting) and smaller for the more localized Co d orbitals (yielding an unresolved a' shoulder on the *a* feature). In the T = Cr case, the broader *d* bands and larger number of t_{2g} holes broaden the threshold features and enhance the a'-feature intensity. Finally, although not germane to this work, it should be noted that a full treatment of such threshold features should include exchange and multiplet effects, particularly for the Cr compound.

Figure 1(b) compares the S K- and Ir L_3 -edge spectra on the same (albeit displaced) energy scale for the I and M phases of CuIr₂S₄. The Ir L_3 edge also manifests an intense WL feature due to the 5d states above E_F . The A feature, at the L_3 edge of CuIr₂S₄, involves the four empty e_g states per Ir, and the aligned S K a peak is associated with the transitions to S p/Ir $d(e_g)$ hybridized states. The $\frac{1}{2}t_{2g}$ hole per Ir makes a weak unresolved contribution to the Ir L_3 WL near the A' energy range; however, the S K a' feature, involving



FIG. 2. Comparison of the S K edges of CuIr_2S_4 just across the *I*-phase (225 K) to *M*-phase (239 K) transition. The difference (bottom) between the *I*- and *M*-phase spectra is shown to highlight the full details of the *M* to *I* electronic state changes.

S $p/\text{Ir } d(t_{2g})$ hybridized states, defines these t_{2g} states much more sharply. The alignment of the Ir $L_3 B$ feature and S K b feature for the CuIr₂S₄ suggests S-Ir hybridization effects at these higher energies.¹²

For comparison, the Ir L_3 edges of IrO₂, CuIr₂Se₄, and Ir metal are also shown in Fig. 1(c). The higher WL intensity and chemical shift of the IrO₂ spectrum reflect its higher valence relative to CuIr₂S₄. The Ir L_3 spectrum of CuIr₂Se₄ shows a greater Ir *d* DOS near E_F , relative to CuIr₂S₄, consistent with a higher density of overlapping states near E_F in the always-metallic Se isomorph.⁶ The combination of monochromator resolution, core hole broadening, and corehole/*d*-electron interactions make the Ir L_3 WL feature essentially identical in the *M* and *I* phases of CuIr₂S₄.

S *K*-edge spectra were taken as the sample slowly warmed through the *I-M* transition (with temperature measurements being ± 5 K) and the details of the discontinuous spectral change can be seen by comparing the $T \sim 225$ K *I*-phase and $T \sim 239$ K *M*-phase spectra in Fig. 2 (top). The difference between these spectra, shown in Fig. 2 (bottom) provides a direct estimate of the detailed Ir d/S-p state redistribution occurring at the transition. Here we will focus only on the central element of this redistribution, the dramatic shift of the $a'(t_{2g})$ feature to higher energy in the *I* phase. A proper theoretical treatment of this transition should also replicate the state redistribution on the low energy side of the *a* feature.

We tacitly assume the S K threshold features are dominated by electronic structure effects in analogy to all past O K-edge threshold measurements.^{14–16} The close quantitative similarity of our *I*-phase S K threshold spectra to the *I*-phase inverse photoemission results⁶ strongly supports this assumption.

We will pattern our proposals for understanding CuIr_2S_4 after those of the Abbate *et al.*¹⁵ reformulation of Goodenough's ideas¹⁸ for the paramagnetic-metal-to-dimerizedinsulator transition in VO₂. These authors^{15,18} developed a simple molecular orbital (MO) theory for VO₂, motivated by METAL-INSULATOR TRANSITION IN CuIr₂S₄: XAS . . .



FIG. 3. (a) The Ir atoms (only) in the cubic spinel CuIr_2S_4 . Note the interleaved and Ir-chain structure and threefold chain intersections at the Ir sites (see center Ir). (b) The *I*-phase Ir chains of CuIr_2S_4 with surrounding atoms omitted for clarity. The isolated circles are Ir^{3+} and the circles, connected by heavy lines, are Ir^{4+} - Ir^{4+} dimers. Class *I* chains (light dotted line) are along the triclinic-(110) direction with the Ir^{4+} - Ir^{4+} - Ir^{3+} - Ir^{4+} interatomic distances being 3.06, 3.59, 3.66, and 3.59 Å and the chain repeat distance being 13.95 Å. Class II chains (light dashed line) are along the (011) direction with the interatomic distances being 3.00, 3.72, 3.55, and 3.66 Å and the chain repeat distance being 13.93 Å.

the appearance of a prominent O K threshold feature in the I phase. The model was based on a strong hybridizationinduced splitting of dimer V d states (d_{\parallel} states) oriented along the chains of edge-sharing VO₆ octahedra in the rutile structure.^{15,18} Abbate *et al.*¹⁵ noted, and Sommers *et al.*¹⁹ emphasized, that Mott-Heitler-London electron correlation effects also contribute to the d_{\parallel} splitting.

The spinel structure of $\operatorname{CuIr}_2 S_4$ is decidedly three dimensional, however, as Fig. 3(a) illustrates, it also contains crisscrossing Ir chains with an Ir-Ir spacing of h=3.48 Å $=a\sqrt{2}/4$, where *a* is the lattice parameter. The chains cross in adjacent planes a/2 apart, and cross-linking chains create three-fold chain intersections at the Ir sites (see the cube center). The cell edge-to-edge chain has a length of 4h=13.92 Å and contains four Ir atoms in the cubic cell. In the metallic phase the Ir^{3.5+} atoms, with a configuration of $t_{2g}^{5.5}$, can be thought of as $[t_{2g}^4][t_{2g}^{1.5}]$, where the former bracket constitutes two filled *d* orbitals and the latter a $\frac{3}{4}$ -filled band for the highest-lying $d(t_{2g})$ orbital.

Space limitations preclude detailed discussion of the complex triclinic (a = 11.95 Å, b = 6.98 Å, c = 11.93 Å, α =91.05°, β =108.47°, and γ =91.03°) *I*-phase structure;¹⁰ however, several crucial points should be noted [see Fig. 3(b)]. All Ir atoms are members of charge-ordered \cdots Ir³⁺Ir⁴⁺-Ir⁴⁺Ir³⁺ \cdots chains with dimerized Ir⁴⁺-Ir⁴⁺ pairs. There are two closely related types of chains (I and II) having unit repeat distances of $\sim 4h$, and extending along approximately orthogonal triclinic cell edge-to-edge directions. Planes of chains in these two directions alternate in the third direction. At staggered chain crossing regions, adjacent Ir^{3+} - Ir^{4+} atoms are still close (in the 3.43–3.56 Å range); however, the orientation of the crucial near- E_F , d-orbital charge lobes, within our model, should be along the chains. This should produce minimal overlap between the filled-shell t_{2g}^6 Ir³⁺ on one chain, and the transverse d lobe on the Ir⁴⁺ on the adjacent chain, leading to near- E_F d bands with quasi-1D character (within a 3D geometrical structure).

The four atom repeat unit in the I-phase chains is com-



FIG. 4. Schematic views of the proposed electronic properties of CuIr_2S_4 . (a) The *I*-phase chain repeat unit with IrS_6 octahedra being viewed from above. The overlapping d_{\parallel}^1 orbital lobes of the Ir^{4+} dimer are shown. The filled d_{\parallel}^2 shell Ir^{3+} sites are represented by circles. (b) The proposed MO electronic structure in the *M* phase. (c) The proposed MO electronic structure, in the *I* phase at the dimerized (left) and undimerized (right) sites. Note only the highest lying t_{2g} d states have been shown explicitly in the diagram.

posed of a $t_{2g}^5 - t_{2g}^5$ dimer, bounded by two filled-orbital t_{2g}^6 sites. After Abbate *et al.*,¹⁵ we denote the last filled *d* orbital along the chain as d_{\parallel} and note that the *I*-phase chain sequence would be $d_{\parallel}^2 d_{\parallel}^1 - d_{\parallel}^1 d_{\parallel}^2$. In the extended-zone scheme, the dispersion curve for the d_{\parallel} band would now have new gaps at $\pi/4h$, $\pi/2h$, and $3\pi/4h$. The $3\pi/4h$ gap falls in the range of the Fermi energy of the $\frac{3}{4}$ filled d_{\parallel} band, and the removal of nested states near E_F should play some role in the *M* to *I* transition. Sommers *et al.*¹⁹ emphasized that electron repulsion/correlation effects were important in VO₂, along with direct d_{\parallel} overlap effects. We believe the correlated singlet dimerization energy is crucial here also, and while the direct *d*-overlap effects should be enhanced and the correlation effects reduced in this 5*d* Ir compound, both are anticipated.

In Fig. 4(a) we show a schematic representation of the dimer-containing chain repeat cell indicating the filled d_{\parallel}^2 sites with circles, and the oriented d_{xy} -type charge cloud¹⁰ at the d_{\parallel}^1 sites. The intersite direct d_{\parallel}^1 - d_{\parallel}^1 overlap of the dimer, across the shared octahedral edge, is emphasized.

In the *M* phase of CuIr₂S₄, the itinerant *d* holes are hopping on and off Ir sites along three-fold cross-linked chains, with the spatial orientation of the d_{xy} charge lobes also fluctuating. The transition to the *I* phase involves several components: a Ir³⁺Ir⁴⁺-Ir⁴⁺Ir³⁺ charge ordering with a concomitant quadrupling of the chain cell to $\sim 4h$, an orbital ordering of the charge lobes at each site into one chain and across the shared edge of an Ir⁴⁺-Ir⁴⁺ dimer, and finally, $d_{\parallel}^{\parallel}-d_{\parallel}^{\parallel}$ hybridization into a spin singlet dimer with correlation effects.

In Fig. 4(b) the MO proposal for the *M* phase of CuIr₂S₄ is shown. The bonding (σ)-antibonding (σ^*) orbitals involve Ir $d(e_g)$ states that point toward the S sites, and induce strongly split hybrid states. The less split bonding (π)- antibonding (π^*) orbitals involve Ir *d* states that point between the S sites, and hybridize more weakly. The highest-lying Ir $d(t_{2g})$ orbital [labeled *d* in Fig. 4(b)] is partially filled at E_F .

In the *I* phase there will be two differing MO combinations, one for the $d_{\parallel}^1 - d_{\parallel}^1$ dimer and one for the d_{\parallel}^2 sites which are shown in Fig. 4(c). At the dimer site, the splitting of the *d* states into a bonding d_{\parallel} and antibonding d_{\parallel}^* pair is dramatic. The fact that both the d_{\parallel}^* and d_{\parallel} states carry hybridized S-*p* states with them, away from E_F , has been emphasized in Fig. 4(c) by the additional broader box accompanying these states. At the Ir³⁺ d_{\parallel}^2 site, the closed d_{\parallel}^2 orbital falls below E_F . Thus in this MO model the *M-I* transition involves; the redistribution of the near- E_F states in the *M* phase, into the *I*-phase dimer site bonding/antibonding $(d_{\parallel}/d_{\parallel}^*)$ states, and into the filled t_{2g} states at the Ir³⁺ site.

Referring back to our S K-edge results in Fig. 2, in the M phase we associate the high-lying MO σ^* states and the near- E_F MO $\pi^* d$ states with the S K a and a' features, respectively. In the I phase, the a feature and σ^* states persist relatively unchanged in both the S-K edge results and MO model. The shift of the a' feature to higher energy in the I phase is associated with the splitting of the MO antibond-

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ing d_{\parallel}^* states to above E_F at the dimer sites. The bonding d_{\parallel} states at the dimer sites, and the filled t_{2g} states at the Ir³⁺ sites, are both pulled below E_F and do not contribute to the S K edge. Thus, this simple starting-point model involves the I phase arising from orbital ordering of the *d*-orbital charge lobes into in-chain d_{\parallel} states, and intrachain charge ordering into Ir⁴⁺-Ir⁴⁺ (d_{\parallel}^1 - d_{\parallel}^1) correlated singlet dimers bounded by Ir³⁺ filled d_{\parallel}^2 orbital sites.

More generally, the underlying electronic origin of the M-I transition in this system appears, at present, unique among 5d row compound. Besides explaining our XAS results, the proposed MO picture appears to explain the photoemission gap formation below E_F . Interestingly, the fact that the band structure calculation predicts a metallic state, even in a tetragonally distorted phase (Ref. 9), strongly suggests that the electrons in the I phase have localized character, despite the common belief that 5d electrons form broad bands. Hence, the MI transition would appear to involve electron localization due to correlation effects.

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