

NAME: _____ SIGNATURE _____

As a student at NJIT I _____, will conduct myself in a professional manner and will comply with the provisions of the NJIT Academic Honor Code. I also understand that I must subscribe to the following pledge on major work submitted for credit as described in the NJIT Academic Honor Code:
On my honor, I pledge that I have not violated the provisions of the NJIT Academic Honor Code.

The exam is closed book and closed notes. Choose the answer that is the closest to your result

$$T(^{\circ}\text{C}) = \frac{5}{9} [T(^{\circ}\text{F}) - 32]; \quad T(^{\circ}\text{F}) = \frac{9}{5} T(^{\circ}\text{C}) + 32; \quad T(\text{K}) = [T(^{\circ}\text{C}) + 273]; \quad 1 \text{ m} = 100 \text{ cm} = 1000 \text{ mm} \quad 1 \text{ hr} = 3600 \text{ s}$$

$$\rho = \frac{m}{V}; \quad P = \frac{F}{A}; \quad p_h = \rho gh; \quad 1 \text{ atm} = 1.013 \times 10^5 \text{ Pa}, \quad F_B = \rho_g V_{im}, \quad A_1 v_1 = A_2 v_2$$

$$Av - \text{volume flow rate} \quad p_1 + \frac{1}{2}\rho v_1^2 + \rho gh_1 = p_2 + \frac{1}{2}\rho v_2^2 + \rho gh_2 \quad \text{flow in horizontal pipe: } p_1 + \frac{1}{2}\rho v_1^2 = p_2 + \frac{1}{2}\rho v_2^2$$

$$L - L_0 = \alpha L_0 (T - T_0) \quad \sigma = Y\alpha (T - T_0) \quad V - V_0 = \beta V_0 (T - T_0); \quad 1 \text{ Liter} = 10^{-3} \text{ m}^3 \quad V_{\text{cube}} = a^3 \quad A_{\text{circle}} = \pi r^2$$

Heat: $Q = mc(T - T_0), \quad Q = mL, \quad c - \text{specific heat} \quad L - \text{latent heat} \quad \text{heat lost} = \text{heat gained}$

$$c_{\text{water}} = 4186 \frac{\text{J}}{\text{kg} \cdot ^{\circ}\text{C}}; \quad L_F = 3.33 \times 10^5 \frac{\text{J}}{\text{kg}}; \quad c_{\text{ice}} = 2100 \frac{\text{J}}{\text{kg} \cdot ^{\circ}\text{C}}$$

$$\sigma = 5.67 \times 10^{-8} \text{ W/m}^2 \text{K}^4 \quad R = 8.313 \text{ J/mol} \cdot \text{K}; \quad Q = kA \frac{T_1 - T_2}{L} t$$

$$\frac{\Delta Q}{\Delta t} = e\sigma A T_1^4 \quad \frac{\Delta Q}{\Delta t} = e\sigma A (T_1^4 - T_2^4) \quad n = \frac{\text{mass}}{\text{molecular} - \text{mass}}$$

Ideal Gas $PV = nRT \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}; \quad N_{\text{av}} = 6.02 \times 10^{23} / \text{mol} \quad T - \text{temp. in kelvins}, \quad \rho = \frac{m}{V},$

Oscillations $x = A \cos(\omega t) \quad v = -\omega A \sin(\omega t) \quad \omega = 2\pi f = \frac{2\pi}{T} \quad F = kx \quad \text{period: } T_{\text{spring}} = 2\pi \sqrt{\frac{m}{k}}; \quad f = \frac{1}{T}$

$$\omega = \sqrt{\frac{k}{m}} \quad v_{\text{max}} = A\omega \quad E = \frac{1}{2}mv^2 + \frac{1}{2}kx^2; \quad E = \frac{1}{2}kA^2; \quad E = \frac{1}{2}m(v_{\text{m}})^2 \quad \text{waves: } v = \lambda f; \quad \text{linear mass } \mu = \frac{m}{L};$$

$$v = \sqrt{\frac{F}{\mu}} \quad \textbf{Sound:} \quad v = 343 \text{ m/s} \quad I_0 = 10^{-12} \text{ W/m}^2$$

$$I = \frac{P}{A} = \frac{P}{4\pi r^2} \quad \beta = 10\text{dB} \log \frac{I}{I_0} \quad \beta_2 - \beta_1 = 10\text{dB} \log \frac{I_2}{I_1} \quad f = f_0 \frac{343 \text{ m/s} \pm v_D}{343 \text{ m/s} \pm v_S}$$

$$\lambda = \frac{2L}{n} \quad f = \frac{v}{2L} n \quad \text{open: } \lambda = \frac{2L}{n} \quad f = \frac{v}{2L} n \quad \text{closed: } \lambda = \frac{4L}{n} \quad f = \frac{v}{4L} n$$

Light: $n = \frac{c}{v} \quad \lambda = \lambda_0/n \quad c = 3 \times 10^8 \text{ m/s} \quad n_1 \sin \theta_1 = n_2 \sin \theta_2 \quad n_1 \sin \theta_{\text{cr}} = n_2 \sin 90^{\circ} \quad d \sin \theta = m\lambda; \quad y = D \frac{m\lambda}{d}$

Electricity: $R = \rho \frac{L}{A}; \quad R = R_0[1 + \alpha(T - T_0)]; \quad V = I^*R; \quad P = \frac{E}{\Delta t} \quad I = \frac{\Delta Q}{\Delta t} = \frac{Ne}{t}; \quad e = 1.6 \times 10^{-19} \text{ C}$

$$P = I^2 R = \frac{V^2}{R} = I^*V; \quad U = CV^2 = Q^2 / C \quad I_{\text{rms}} = \frac{V_{\text{rms}}}{R} \quad P_{\text{av}} = V_{\text{rms}} * I_{\text{rms}} = I_{\text{rms}}^2 R = \frac{V_{\text{rms}}^2}{R} \quad F = \kappa \frac{q_1 \cdot q_2}{r_{12}^2}$$

$$R_{\text{par}} = \left(\frac{1}{R_1} + \frac{1}{R_2} + \dots \right)^{-1} \quad R_{\text{series}} = R_1 + R_2 + \dots \quad C_{\text{series}} = \left(\frac{1}{C_1} + \frac{1}{C_2} + \dots \right)^{-1} \quad C_{\text{par}} = C_1 + C_2 + \dots$$