

# MATHEMATICAL & PHYSICAL CONCEPTS IN QUANTUM MECHANICS

## *Operators*

An operator is a symbol which defines the mathematical operation to be carried out on a function.

Examples of operators:

$d/dx$  = first derivative with respect to  $x$

$\sqrt{\quad}$  = take the square root of

$3$  = multiply by 3

Operations with operators:

If  $A$  &  $B$  are operators &  $f$  is a function, then

$$(A + B) f = Af + Bf$$

$$A = d/dx, B = 3, f = x^2$$

$$(d/dx + 3) x^2 = dx^2/dx + 3x^2 = 2x + 3 x^2$$

$$ABf = A (Bf)$$

$$d/dx (3 x^2) = 6x$$

Note that  $A(Bf)$  is not necessarily equal to  $B(Af)$ :

$$A = d/dx, B = x, f = x^2$$

$$A (Bf) = d/dx(x \cdot x^2) = d/dx (x^3) = 3 x^2$$

$$B (Af) = x (d/dx x^2) = 2 x^2$$

$$\text{In general, } d/dx (xf) = f + x df/dx = (1 + x d/dx)f$$

$$\text{So } d/dx x = 1 + x d/dx$$

Since A & B are operators rather than numbers, they don't necessarily *commute*. If two operators A & B commute, then

$$AB = BA$$

and their commutator = 0:

$$[A,B] = AB - BA = 0$$

(Numbers always commute:  $2 \cdot 3 f = 3 \cdot 2 f$ ;  $[2,3] = 0$ )

What is the commutator of  $d/dx$  &  $x$ ?

$$[d/dx, x] = ?$$

Since we have shown that  $d/dx x = 1 + x d/dx$ , then

$$[d/dx, x] = d/dx x - x d/dx = 1$$

What is the commutator of 3 &  $d/dx$ ?

$$[3, d/dx] f = 3 d/dx f - d/dx 3 f = 3 d/dx f - 3 d/dx f = 0 =$$

$$[d/dx, 3]$$

Equality of operators: If  $Af = Bf$ , then  $A = B$

Associative Law:  $A(BC) = (AB)C$

Square of an operator: Apply the operator twice  $A^2 = A A$

$$(d/dx)^2 = d/dx d/dx = d^2/dx^2$$

C = take the complex conjugate;  $f = e^{ix}$

$$C f = (e^{ix})^* = e^{-ix}$$

$$C^2 f = C (Cf) = C (e^{-ix}) = (e^{-ix})^* = e^{ix} = f$$

$$\text{If } C^2 f = f, \text{ then } C^2 = 1$$

Linear Operator: A is a linear operator if

$$A(f + g) = Af + Ag$$

$$A(cf) = c (Af)$$

where  $f$  &  $g$  are functions &  $c$  is a constant.

Examples of linear operators:

$$d/dx (f + g) = df/dx + dg/dx$$

$$3(f + g) = 3f + 3g$$

Examples of nonlinear operators:

$$\sqrt{f + g} \text{ is not equal to } \sqrt{f} + \sqrt{g}$$

$$\text{inverse } (f + g) = 1/(f + g) \text{ is not equal to } 1/f + 1/g$$

Cautionary note: When trying to determine the result of operations with operators that include partial derivatives, always

using a function as a “place holder”. For example, what is  $(d/dx + x)^2$ ?

$$\begin{aligned}(d/dx + x)^2 f &= (d/dx + x) (d/dx + x) f \\ &= (d/dx + x) (df/dx + xf) \\ &= d/dx (df/dx + xf) + x (df/dx + xf) \\ &= d^2 f/dx^2 + d/dx (xf) + x (df/dx) + x^2 f \\ &= d^2 f/dx^2 + x df/dx + f + x (df/dx) + x^2 f \\ &= (d^2/dx^2 + 2x d/dx + 1 + x^2) f\end{aligned}$$

$$\text{So } (d/dx + x)^2 = (d^2/dx^2 + 2x d/dx + 1 + x^2)$$

### *Eigenfunction/Eigenvalue Relationship:*

When an operator operating on a function results in a constant times the function, the function is called an eigenfunction of the operator & the constant is called the eigenvalue

$$A f(x) = k f(x)$$

$f(x)$  is the eigenfunction &  $k$  is the eigenvalue

$$\text{Example: } d/dx(e^{2x}) = 2 e^{2x}$$

$e^{2x}$  is the eigenfunction; 2 is the eigenvalue

How many different eigenfunctions are there for the operator  $d/dx$ ?

$$df(x)/dx = k f(x)$$

Rearrange the eq. to give:  $df(x)/f(x) = k dx$

and integrate both sides:  $\int df(x)/f(x) = \int k dx$

to give:  $\ln f = kx + C$

$$f = e^{kx+C} = e^{kx} e^C = e^{kx} C', C' = e^C$$

Since there are no restrictions on  $k$ , there are an infinite number of eigenfunctions of  $d/dx$  of this form.

$C'$  is an arbitrary constant. Each choice of  $k$  leads to a different solution. Each choice of  $C'$  leads to multiples of the same solution.

Any eigenfunction of a linear operator can be multiplied by a constant and still be an eigenfunction of the operator. This means that if  $f(x)$  is an eigenfunction of  $A$  with eigenvalue  $k$ , then  $cf(x)$  is also an eigenfunction of  $A$  with eigenvalue  $k$ . Prove it:

$$A f(x) = k f(x)$$

$$A [cf(x)] = c [Af(x)] = c [kf(x)] = k [cf(x)]$$

To specify the type of eigenfunction of  $d/dx$  more definitively, one can apply a physical constraint on the eigenfunction, as we did with the Particle in a Box:

$$c e^{kx} \text{ must be finite as } x \rightarrow \pm\infty$$

The most general  $k$  is a complex number:  $k = a + ib$

$$\text{Then } c e^{kx} = c e^{(a+ib)x} = c e^{ax} e^{ibx} = c e^{ax} (\cos bx + i \sin bx)$$

Since  $e^{ax} \rightarrow \infty$  for  $x \rightarrow \pm\infty$ ,  $a$  must be 0

$b$  can be any number

So  $c e^{ibx}$  is the correct eigenfunction of  $d/dx$ .

### *Relationship of Quantum Mechanical Operators to Classical Mechanical Operators*

In the 1-dimensional Schrödinger Eq.

$$[(-\hbar^2/2m) d^2/dx^2 + V(x)] \psi(x) = E \psi(x),$$

$\psi(x)$  is the eigenfunction,  $E$  is the eigenvalue, & the Hamiltonian operator is

$$(-\hbar^2/2m) d^2/dx^2 + V(x)$$

The Hamiltonian function was originally defined in classical mechanics for systems where the total energy was conserved. This occurs when the potential energy is a function of the coordinates only. This is the type of system to be studied with quantum mechanics.

The classical Hamiltonian expressed Newton's Eq. of Motion such that the energy was a function of the coordinates  $(x,y,z)$  & conjugate momentum  $(p_x, p_y, p_z)$  where

$$p_x = m v_x \qquad v_x = p_x / m$$

with  $m =$  mass &  $v_x =$  velocity in the  $x$ -direction

Classical kinetic energy (KE) is defined as

$$KE_x = (1/2) m v_x^2 = p_x^2/(2m)$$

The classical Hamiltonian function is defined as the sum of the kinetic energy (a function of momentum) & the potential energy (a function of coordinates)

$$H = p_x^2/(2m) + V(x)$$

for a 1-dimensional system

Comparison to the Schrödinger Eq. shows that

$$(-\hbar^2/2m) d^2/dx^2 \leftrightarrow p_x^2/(2m)$$

*Some Postulates of Quantum Mechanics:*

(1) Postulate: For every physical property, there is a quantum mechanical operator

(2) Postulate: To find the operator, write the classical mechanical expression for the property

$$F(x,y,z,p_x, p_y, p_z)$$

then substitute as follows:

Each coordinate operator,  $q$ , is replaced by multiplication by the coordinate

$$\text{operator } q = q \cdot \qquad q=x,y,z$$

Each Cartesian component of momentum ( $p_x, p_y, p_z$ ) is replaced by the operator

$$p_q = (\hbar/i) \partial/\partial q = -i \hbar \partial/\partial q, \qquad q=x,y,z$$

So operator  $x = x \cdot$ , etc. ,  $p_x = -i \hbar \partial/\partial x$ , etc.

$$\text{Then } p_x^2 = (-i \hbar \partial/\partial x)^2 = (i)^2 \hbar^2 \partial^2/\partial x^2 = - \hbar^2 \partial^2/\partial x^2$$

Potential energy functions are usually functions of the coordinates, such as

$$V(x) = a x^2$$

In general, the operator  $V(x)$  is replaced by multiplication by  $V(x)$ :  $V(x) \cdot$

In summary

Classical mechanics (1-dimension)

$$H = T + V = \text{KE} + \text{PE} = p_x^2/(2m) + V(x)$$

Quantum mechanics (1-dimension)

$$H (\text{operator}) = T (\text{operator}) + V (\text{operator})$$

$$= - (\hbar^2/2m) d^2/dx^2 + V(x)$$

(3) Postulate: The eigenvalues of a system are the only value a property can have

$$H = \text{Hamiltonian energy operator} = - (\hbar^2/2m) d^2/dx^2 + V(x)$$

$$H \psi_i = E_i \psi_i \quad i=1,2,.. \text{ different states}$$

Measurement of the energy of the system will result in one of the  $E_i$  (eigenvalues, observables)



Example: Is  $\Psi(x,t)$  an eigenfunction of the  $p_x$  operator for the 1-dimensional particle in a box?

$$\Psi(x,t) = e^{iEt/\hbar} \psi(x) \quad \text{state function}$$

$$\psi(x) = \sqrt{2/L} \sin(n\pi x/L), \quad E_n = n^2 \hbar^2 / (8mL^2)$$

$$p_x = -i \hbar \partial / \partial x$$

For  $\Psi(x,t)$  to be an eigenfunction of  $p_x$ , must have

$$p_x \Psi(x,t) = c \Psi(x,t)$$

But  $d/dx \sin(Ax) = A \cos(Ax)$ , so  $\Psi(x,t)$  is not an eigenfunction of  $p_x$

Example: Is  $\Psi(x,t)$  an eigenfunction of the  $p_x^2$  operator for the 1-dimensional particle in a box?

$$\begin{aligned} p_x^2 \Psi(x,t) &= -\hbar^2 (d^2/dx^2) \{ e^{iEt/\hbar} \sqrt{2/L} \sin(n\pi x/L) \} \\ &= -\hbar^2 e^{iEt/\hbar} \sqrt{2/L} (n\pi/L) d/dx \cos(n\pi x/L) \\ &= \hbar^2 e^{iEt/\hbar} \sqrt{2/L} (n\pi/L)^2 \sin(n\pi x/L) \\ &= \hbar^2 (n\pi/L)^2 \{ e^{iEt/\hbar} \sqrt{2/L} \sin(n\pi x/L) \} \\ &= \hbar^2 (n\pi/L)^2 \Psi(x,t) \\ &= \hbar^2 (n^2/(4L^2)) \Psi(x,t) \quad \text{Yes} \end{aligned}$$

Since  $n=1,2,\dots$ , the eigenvalue  $\hbar^2 (n^2/(4L^2))$  is quantized.

Find the eigenfunctions of  $p_x$ .

$$p_x g(x) = k g(x)$$

$$-i\hbar \frac{dg}{dx} = k g$$

$$\frac{dg}{g} = (ik/\hbar) dx$$

$$\ln g = (ik/\hbar)x + C$$

$$g = A e^{(ik/\hbar)x}$$

To keep  $g$  well-behaved as  $x \rightarrow \pm \infty$ ,  $k$  must be real. So the eigenvalues of  $p_x$  are all the real numbers  $k$ ,  $-\infty < k < \infty$ .

### *Forms of Operators in 3-Dimensions & More Than 1 Particle*

#### One particle in 3-dimensions:

$$T = (-\hbar^2/2m) (\partial^2/\partial x^2 + \partial^2/\partial y^2 + \partial^2/\partial z^2)$$

$$= (-\hbar^2/2m) \nabla^2 \quad \nabla^2 \text{ is the Laplacian operator}$$

$$H \psi(x,y,z) = \{(-\hbar^2/2m)\nabla^2 + V(x,y,z)\} \psi(x,y,z) = E \psi(x,y,z)$$

The probability of finding the particle at time  $t$  in a region bounded by  $(x,y,z)$  &  $(x+dx,y+dy,z+dz)$  is

$$|\psi(x,y,z,t)|^2 dx dy dz \quad d\tau = dx dy dz$$

$$1 = \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} |\psi(x,y,z,t)|^2 d\tau$$

#### n particles in 3-dimensions:

Particle  $i$  has mass  $m_i$ , position  $(x_i, y_i, z_i)$  and momentum  $(p_{xi}, p_{yi}, p_{zi})$

$$\begin{aligned}
T &= (-\hbar^2/2m_1) (\partial^2/\partial x_1^2 + \partial^2/\partial y_1^2 + \partial^2/\partial z_1^2) + \\
&\quad (-\hbar^2/2m_2) (\partial^2/\partial x_2^2 + \partial^2/\partial y_2^2 + \partial^2/\partial z_2^2) + \dots + \\
&\quad (-\hbar^2/2m_n) (\partial^2/\partial x_n^2 + \partial^2/\partial y_n^2 + \partial^2/\partial z_n^2) \\
&= \sum_{i=1}^n (-\hbar^2/2m_i) \nabla_i^2
\end{aligned}$$

If  $V$  depends only on the Cartesian coordinates,

$$V = V(x_1, y_1, z_1, \dots, x_n, y_n, z_n)$$

Then  $\psi = \psi(x_1, y_1, z_1, \dots, x_n, y_n, z_n)$  and

$$H \psi = \left\{ \sum_{i=1}^n (-\hbar^2/2m_i) \nabla_i^2 + V(x_1, \dots, z_n) \right\} \psi = E \psi$$

The probability of finding the first particle in a region bounded by  $(x_1, y_1, z_1)$  &  $(x_1+dx_1, y_1+dy_1, z_1+dz_1)$ , the second particle in a region bounded by  $(x_2, y_2, z_2)$  &  $(x_2+dx_2, y_2+dy_2, z_2+dz_2)$ , etc. is

$$|\psi(x_1, y_1, z_1, \dots, x_n, y_n, z_n, t)|^2 d\tau$$

$$d\tau = dx_1 dy_1 dz_1 \dots dx_n dy_n dz_n$$

$$1 = \int_{-\infty}^{\infty} \dots \int_{-\infty}^{\infty} |\psi(x_1, y_1, z_1, \dots, x_n, y_n, z_n, t)|^2 d\tau$$

*Particle in a 3-Dimensional Box:*

$$V(x, y, z) = 0 \quad 0 < x < a, \quad 0 < y < b, \quad 0 < z < c$$

$\psi = 0$  outside the box, as in the 1-dimensional case

Inside the box:  $H \psi = E \psi$

$$(-\hbar^2/2m) (\partial^2/\partial x^2 + \partial^2/\partial y^2 + \partial^2/\partial z^2) = E \psi$$

Solve by Method of Separation of Variables: Assume that  $\psi$  is a product of functions, each depending only on one variable. This is a reasonable assumption because the potential has no cross terms (i.e. terms including products of variables)

$$\psi(x,y,z) = f(x) g(y) h(z)$$

$$H \psi = (-\hbar^2/2m) \{g h d^2f/dx^2 + f h d^2g/dy^2 + f g d^2h/dz^2\} = E f(x) g(y) h(z)$$

Dividing both sides by  $f(x) g(y) h(z)$  gives:

$$(-\hbar^2/2m) \{ (1/f) d^2f/dx^2 + (1/g) d^2g/dy^2 + (1/h) d^2h/dz^2 \} = E$$

Can rewrite so that the left-hand side depends only on  $x$  & the right-hand side depends only on  $y$  &  $z$ :

$$(1/f) d^2f/dx^2 = - (1/g) d^2g/dy^2 - (1/h) d^2h/dz^2 - 2mE/\hbar^2$$

But this means that the left & right-hand sides must be equal to a constant.

$$\text{Let } k_x = (1/f) d^2f/dx^2$$

Could rewrite the eq. so that the left-hand side depends only on  $y$ , etc. and get

$$k_y = (1/g) d^2g/dy^2$$

$$k_z = (1/h) d^2h/dz^2$$

$$\text{with } k_x + k_y + k_z = -2mE/\hbar^2$$

Can redefine the energy components as

$$k_x = -2mE_x/\hbar^2, \text{ etc.}$$

$$\text{So that } E_x + E_y + E_z = E$$

$$\text{and } (1/f) d^2f/dx^2 = -2mE_x/\hbar^2, \text{ etc.}$$

$$\text{Then } d^2f/dx^2 + 2mE_x/\hbar^2 f = 0$$

$$d^2g/dy^2 + 2mE_y/\hbar^2 g = 0$$

$$d^2h/dz^2 + 2mE_z/\hbar^2 h = 0$$

Boundary Conditions: Functions must be zero at the walls.

$$f(x) = 0 \text{ at } x = 0, a$$

$$g(y) = 0 \text{ at } y = 0, b$$

$$h(z) = 0 \text{ at } z = 0, c$$

So the solutions are the same as for the 1-dimensional particle in a box:

$$f(x) = \sqrt{2/a} \sin(n_x \pi x/a), E_x = (n_x^2 \hbar^2)/(8ma^2), n_x=1,2,\dots$$

$$g(y) = \sqrt{2/b} \sin(n_y \pi y/b), E_y = (n_y^2 \hbar^2)/(8mb^2), n_y=1,2,\dots$$

$$h(z) = \sqrt{2/c} \sin(n_z \pi z/c), E_z = (n_z^2 \hbar^2)/(8mc^2), n_z=1,2,\dots$$

$$E = E_x + E_y + E_z = (\hbar^2)/(8m) \{n_x^2/a^2 + n_y^2/b^2 + n_z^2/c^2\}$$

with the quantum numbers  $n_x, n_y, n_z$  varying independently

$$\psi(x,y,z) = \sqrt{[8/(abc)]} \sin(n_x \pi x/a) \sin(n_y \pi y/b) \sin(n_z \pi z/c)$$

Normalize  $\psi$ :

$$\begin{aligned} 1 &= \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} |\psi(x,y,z,t)|^2 d\tau \\ &= \int_0^a dx |f(x)|^2 \int_0^b dy |g(y)|^2 \int_0^c dz |h(z)|^2 \end{aligned}$$

But each function is separately normalized

$$1 = \int_0^a dx |f(x)|^2, \text{ etc.}$$

so  $\psi$  is automatically normalized.

Consider a particle in a cube:  $a = b = c$ ,

$$E = (h^2)/(8m a^2) \{n_x^2 + n_y^2 + n_z^2\}$$

$$\text{or } \{n_x^2 + n_y^2 + n_z^2\} = (E 8m a^2)/ h^2$$

Tabulate

$n_x$	$n_y$	$n_z$	111	211	121	112	122	221	212
$\{n_x^2 + n_y^2 + n_z^2\}$			3	6	6	6	9	9	9

$n_x$	$n_y$	$n_z$	113	131	311	222	etc
$\{n_x^2 + n_y^2 + n_z^2\}$			11	11	11	12	

*Degeneracy* occurs when two or more independent wavefunctions correspond to states with the same energy eigenvalue

Each set of  $(n_x, n_y, n_z)$  corresponds to an independent wavefunction. Since there are 3 independent wavefunctions which give  $\{n_x^2 + n_y^2 + n_z^2\} = 6$ , the corresponding energy level is said to be 3-fold degenerate.

A rectangular box wouldn't have degenerate energy levels. Degeneracy is related to the symmetry of the system.

The *degree of degeneracy* of an energy level equals the number of linearly independent wavefunctions corresponding to that value of the energy.

A set of  $n$  functions is said to be *linearly independent* if no member of the set can be written as a linear combination of the others.

$\psi_1, \psi_2, \psi_3$ , etc are linearly independent if

$$c_1\psi_1 + c_2\psi_2 + \dots + c_n\psi_n = 0 \text{ only if } c_1 = c_2 = \dots = c_n = 0$$

Example:  $f_1 = 3x, f_2 = 5x^2 - x, f_3 = x^2$

$$f_2 = 5 f_3 - f_1/3 \quad \text{not linearly independent}$$

Example:  $f_1 = 1, f_2 = x, f_3 = x^2$       linearly independent

Theorem: For any set of linearly independent eigenfunctions of the Hamiltonian operator,  $(\psi_1, \psi_2, \dots, \psi_n)$ , with eigenvalue  $\omega$ , any linear combination of these eigenfunctions is also an eigenfunction of  $H$  with eigenvalue  $\omega$ .

Prove that for

If 
$$\phi = c_1 \psi_1 + c_2 \psi_2 + \dots + c_n \psi_n ,$$

and  $H \psi_i = \omega \psi_i$  for  $i = 1, \dots, n$

then  $H \phi = \omega \phi$

Proof: 
$$\begin{aligned} H \phi &= H (c_1 \psi_1 + c_2 \psi_2 + \dots + c_n \psi_n) \\ &= c_1 H \psi_1 + c_2 H \psi_2 + \dots + c_n H \psi_n \\ &= c_1 \omega \psi_1 + c_2 \omega \psi_2 + \dots + c_n \omega \psi_n \\ &= \omega (c_1 \psi_1 + c_2 \psi_2 + \dots + c_n \psi_n) \\ &= \omega \phi \end{aligned}$$

Note that the degree of degeneracy of energy level  $\omega$  is the number of linearly independent eigenfunctions ( $n$ ) belonging to that level.

*Average (or Expectation) Value of a Physical Property:*

For a quantity that depends on *discrete* changes in the variables, the average value is defined by a *sum*

F - the physical property

$\langle F \rangle$  - average value of F

N - the number of systems that are measured

$f_i$  - an observed value of F

$n_f$  - the number of times f is observed

f - a possible value of F

$$\langle F \rangle = \sum_{i=1}^N f_i / N = \sum_{i=1}^N n_f f / N$$

Example: In a class there are 9 ( $N=9$ ) students. On a quiz the grades are: 0 ( $f_1$ ), 20 ( $f_2$ ), 20 ( $f_3$ ), 60 ( $f_4$ ), 60 ( $f_5$ ), 80



( $f_6$ ), 80 ( $f_7$ ), 80 ( $f_8$ ), 100 ( $f_9$ ). There are 5 questions & each question is either all right (20 points) or all wrong (0 points). Calculate the average grade.

$$\langle F \rangle = \sum_{i=1}^N f_i / N = (1/9) [0 + 20 + 20 + 60 + 60 + 80 + 80 + 80 + 100] = 56$$

Alternatively,

$$\langle F \rangle = \sum n_f f / N$$

The  $f$  possible values of  $F$  (and  $n_f$  number of times  $f$  is observed) are:

0 (1), 20 (2), 40 (0), 60 (2), 80 (3), 100 (1)

$$\langle F \rangle = (1/9) [1 \cdot 0 + 2 \cdot 20 + 0 \cdot 40 + 2 \cdot 60 + 3 \cdot 80 + 1 \cdot 100] = 56$$

Note that the average grade is not one of the possible or observed grades.

Since the probability,  $P_f$ , is defined as  $n_f / N$ , then  $\langle F \rangle$  can be written as

$$\langle F \rangle = \sum_f P_f f$$

For quantities that depend on variables that can take on a continuous range of values,

$$P_f = |\psi|^2 d\tau \qquad \sum_f \rightarrow \int$$

$$\langle F \rangle = \int \Psi^* F \Psi \, d\tau,$$

where  $\Psi$  is the time-dependent wavefunction

Since  $F$  is an operator, cannot write  $|\Psi|^2 F$ .

Must have  $\Psi^* F \Psi$ , unless  $F$  is a function of coordinates only

$\int d\tau$  is shorthand notation which means integrate over the correct variables & volume element.

For  $n$  particles in 3 dimensions,  $\int d\tau =$

$$\int_{-\infty}^{\infty} dx_1 \int_{-\infty}^{\infty} dy_1 \int_{-\infty}^{\infty} dz_1 \dots \int_{-\infty}^{\infty} dx_n \int_{-\infty}^{\infty} dy_n \int_{-\infty}^{\infty} dz_n$$

For 1 particle in 1 dimension,  $\int d\tau = \int_{-\infty}^{\infty} dx$

A stationary state is defined as one for which the probability density doesn't vary in time

$$d |\Psi|^2 / dt = 0$$

For these states (& if  $F$  is independent of time), one can show that

$$\Psi^* F \Psi = \psi^* F \psi.$$

This is because

$$\Psi = e^{-iEt/\hbar} \psi$$

$$\text{So } \Psi^* F \Psi = e^{iEt/\hbar} \psi^* F e^{-iEt/\hbar} \psi$$

Since  $F$  is independent of time,  $F e^{-iEt/\hbar} \psi = e^{-iEt/\hbar} F \psi$ ,

and  $\Psi^* F \Psi = e^{iEt/\hbar} e^{-iEt/\hbar} \psi^* F \psi = \psi^* F \psi$ .

The average value of a sum of operators equals the sum of the average values of the operators:

$$\langle F + G \rangle = \langle F \rangle + \langle G \rangle$$

But the average value of a product of operators is *not* equal to the product of the average values of the operators:

$$\langle F \cdot G \rangle \text{ is not equal to } \langle F \rangle \cdot \langle G \rangle$$

Example: Find  $\langle F \rangle$  for  $F \psi = k \psi$ .

$$\begin{aligned} \langle F \rangle &= \int \psi^* F \psi \, d\tau = \int \psi^* k \psi \, d\tau = k \int \psi^* \psi \, d\tau \\ &= k \end{aligned}$$

since  $\int \psi^* \psi \, d\tau = 1$ .