PROBLEM SET SOLUTIONS

Chapter 3, Quantum Chemistry, 5th Ed., Levine

3.23Which of the following functions are eigenfunctions of d^2/dx^2 ?

To be an eigenfunction of d^2/dx^2 , the function must satisfy the following relationship, where k is a constant:

$$d^2/dx^2$$
 $\psi = k \psi$.

(A) d^2/dx^2 (e^x) = d/dx (e^x) = e^x. Yes it is an eigenfunction with eigenvalue = 1.

(B)
$$d^2/dx^2$$
 (x^2) = d/dx ($2x$) = 2. No

3.27 Evaluate the commutators:

(A)
$$[x, p_x] \psi = x p_x \psi - p_x (x \psi) = x (-i\underline{h} d/dx) \psi - (-i\underline{h} d/dx) (x \psi)$$

$$= (-i\underline{h}) (x d\psi/dx - (1 \cdot \psi + x d\psi/dx)) = (i\underline{h}) \psi$$
 $[x, p_x] = i\underline{h}$

(B)
$$[x, p_x^2] \psi = x p_x^2 \psi - p_x^2 (x \psi) = x (-\underline{h}^2 d^2 / dx^2) \psi - (-\underline{h}^2 d^2 / dx^2) (x \psi)$$

 $= (-\underline{h}^2) (x d^2 \psi / dx^2 - d / dx (x d\psi / dx + \psi \cdot 1)$
 $= (-\underline{h})^2 (x d^2 \psi / dx^2 - (d\psi / dx + x d^2 \psi / dx^2 + d\psi / dx))$
 $= (-\underline{h}^2) (-2 d\psi / dx) = 2 \underline{h}^2 d\psi / dx$
 $[x, p_x^2] = 2 \underline{h}^2 d / dx$

3.36 The terms *state* and *energy* level are not synonymous in quantum mechanics. For the particle in a cubic box, consider the energy range $E < 15 \text{ h}^2/(8\text{ma}^2)$.

(A) How many states lie in this range?

(B) How many energy levels lie in this range?

For a particle in a cubic box of length a on a side,

$$E = (n_x^2 + n_y^2 + n_z^2) h^2/(8ma^2).$$

Rearranging gives

$$E (8ma^2)/h^2 = (n_x^2 + n_y^2 + n_z^2)$$

Rephrase the question: How many states lie in the range $(n_x^2 + n_y^2 + n_z^2) < 15$?

$n_x, n_y, n_z,$	1,1,	2,1,	1,2,	1,1,	1,2,	2,1,	2,2,	1,1,	1,3,	3,1,
	1	1	1	2	2	2	1	3	1	1
$E(8ma^2)/h^2$	3	6	6	6	9	9	9	11	11	11

n_x , n_y , n_z ,	2,2,	2,1,	1,2,	3,2,	2,3,	1,3,	3,1,	2,2,	2,3,	3,2,
	2	3	3	1	1	2	2	3	2	2
$E(8ma^2)/h^2$	12	14	14	14	14	14	14	17	17	17

n_x , n_y , n_z ,	1,1,	4,1,	1,4,	1,3,	3,1,	3,3,
	4	1	1	3	3	1
$E(8ma^2)/h^2$	18	18	18	19	19	19

TOTAL: 17 states and 6 energy levels lie within this range.

For example, one level has 6 states with E $(8ma^2)/h^2 = 14$

3.39For the particle confined to a box with dimensions a, b, and c, find the following values for the state with quantum numbers n_x , n_y , n_z . (NOTE: <x> & <p_x> are done in Example, p. 56)

For the particle in a 3D box,

$$E = (n_x^2/a^2 + n_y^2/b^2 + n_z^2/c^2) (h/8m);$$

 $\psi(x,y,z) = \sqrt{(8/(abc))} \sin(n_x \pi x/a) \sin(n_y \pi y/b) \sin(n_z \pi z/c) = f(x) g(y) h(z); d\tau = dx dy dz$ $\int_0^a f^*(x) f(x) dx = 1 = \int_0^b g^*(y) g(y) dy = \int_0^c h^*(z) h(z) dz$

- (A) <x> = $\int \psi^* x \psi \, d\tau = \int_0^a f^*(x) x f(x) dx \int_0^b g^*(y) g(y) dy \int_0^c h^*(z) h(z) dz$ = $\int_0^a x (2/a) \sin^2 (n_x \pi x/a) dx = a/2$ (For ground state, $n_x = 1$)
- (B) $\langle y \rangle \& \langle z \rangle$ have exactly the same mathematical form as $\langle x \rangle$ so $\langle y \rangle = b/2 \& z = c/2$.

(D)
$$\langle x^2 \rangle = \int \psi^* x^2 \psi \, d\tau = \int_0^a f^*(x) \, x^2 \, f(x) \, dx \, \int_0^b g^*(y) \, g(y) \, dy \, \int_0^c h^*(z) \, h(z) \, dz$$

= $(2/a) \int_0^a x^2 \sin^2(n_x \pi x/a) \, dx$

$$= (2/a) (x^3/6 - (x^2a/(4n_x\pi) - a^3/(8n_x^2\pi^2)) \sin (2n_x\pi x/a) - xa^2/(4n_x^2\pi^2) \cos (2n_x\pi x/a) \int_0^a (A4)^{-1} dx dx$$

$$=(2/a)\;(a^3/6 - (a^3/(4n_x\pi) - a^3/(8n_x^{\ 2}\pi^2))\;sin\;(2n_x\pi) - a^3/(4n_x^{\ 2}\pi^2)\;cos\;(2n_x\pi)) - 0 - 0)$$

=
$$(2/a) (a^3/6 - (a^3/(4n_x\pi) - a^3/(8n_x^2\pi^2)) \cdot 0 - a^3/(4n_x^2\pi^2) \cdot 1)$$

=
$$(2/a) (a^3/6 - a^3/(4n_x^2\pi^2))$$

=
$$a^2/3 - a^2/(2n_x^2\pi^2) = (1 - 3/(2n_x^2\pi^2)) a^2/3$$

Does $\langle x^2 \rangle = \langle x \rangle^2$? No because $\langle x \rangle = a/2 \& \langle x \rangle^2 = a^2/4$.

Does
$$< xy > = < x > < y > ?$$

Yes, because $\langle xy \rangle = (8/(abc)) (a^2/4) (b^2/4) (c^2/4) = ab/4 = (a/2) (b/2) = \langle x \rangle \langle y \rangle$.